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Check: Does Answer Make Any Sense? Note: equilibrium concentrations must be in Molarity, moles/liter. If information is given in grams or moles plus solvent volume, you will need to convert into molarity. Use the ICE method to find the equilibrium concentrations and the K values for the following: C —OLIO 0.10 000 C 21) +0dC oos coo 0.20

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The mole concept provides this bridge, and is central to all of quantitative chemistry. Counting Atoms: Avogadro's Number. Owing to their tiny size, atoms and molecules cannot be counted by direct observation. But much as we do when "counting" beans in a jar, we can estimate the number of particles in a sample of an element or compound if we ...

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